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## CBSE Class 10 Science Lab Manual

CBSE Class 10 Science Lab Manual – pH of Samples Dilute hydrochloric acid (HCl). Dilute ethanoic acid (CH<sub>3</sub>COOH) solution. Lemon juice. Water (H<sub>2</sub>O). Dilute sodium bicarbonate (NaHCO<sub>3</sub>) solution.

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NCERT Class 10 Science Lab Manual – pH of Samples. This is because at higher temperature the water will

dissociate more into ions. > If the temperature of water decreases its pH is more than 7 but the water remains neutral. This is because at lower temperature the water will dissociate less into ions.

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